

Point Defects in CVD-grown Monolayer WSe₂ studied by STM/STS and DFT Modeling

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Motivation

- ◆ Single layer (SL) WSe₂ has potential applications in nanoelectronics, gas-sensing, and other applications.
- ◆ Defects can change the electronic and chemical properties of any material.
- ◆ Important to understand defects in WSe₂ at the atomic scale.

Methods

- ◆ **Growth:** SL-WSe₂ was grown by chemical vapor deposition (CVD)
- ◆ **Atomic-scale Imaging:** Scanning Tunneling Microscopy/Scanning Tunneling Spectroscopy
 - Operated at ~77 K under ultrahigh vacuum conditions (10⁻¹⁰ mbar).
 - STS acquired by a lock-in amplifier with a sinusoidal modulation of 40 mV at 625 Hz.
 - Bias voltage (V_{Tip}) is applied on the STM tip respect to the sample.
 - Electrochemically etched tungsten tips were used in all measurements.
- ◆ **Modeling:** Density Functional Theory (DFT) Calculations
 - VASP code, PBE exchange-correlation functional with Grimme's correction for van der Waals interactions
 - Energy cutoff: 400 eV
 - Force convergence criterion: 0.01 eV/Å for graphite, bulk WSe₂ and SL-WSe₂ supercell and 0.05 eV/Å for WSe₂/graphite supercell
 - Tersoff-Hamann approximation for simulation of STM images
- ◆ **Workflow**
 - CVD-grown SL-WSe₂ was deposited on graphite substrates
 - STM and STS were performed
 - DFT calculations on many atomic models were performed, including vacancies, antisite defects, intercalation or adsorption of W, Se and O
 - STM images and densities of states were simulated to compare with experiment
 - Only atomic structures involving O substitution, O insertion, and O adsorption were consistent with experiment

Conclusion

- ◆ Point defects have been observed in atomic-scale STM images of CVD-grown SL-WSe₂.
- ◆ Extensive DFT modeling of different structures strongly suggests that these point defects are all related to the presence of atomic oxygen (O substitution, O insertion and O adsorption)
- ◆ It is likely that these defects are due to the dissociation of molecular oxygen on Se vacancy sites
- ◆ These defects do not have a large impact on electronic structure close to the valence and conduction band edges (their STS and PDOS are similar to pristine WSe₂).
- ◆ However, O substitution and O insertion defects increase the chemical activity of WSe₂ (not shown).

Acknowledgements

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Results

WSe₂ on graphite

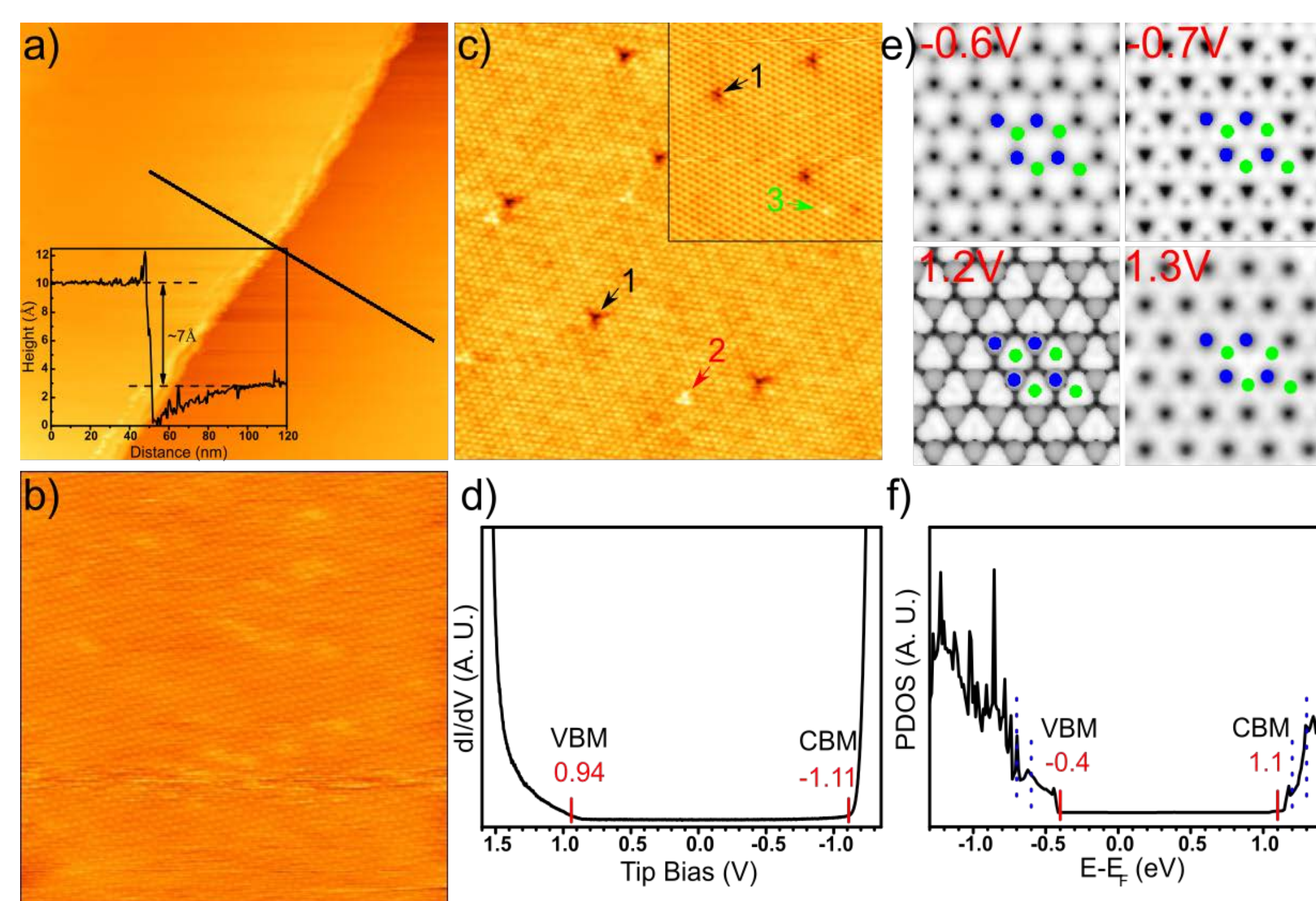


Figure 1. **a)** ($V_{\text{Tip}} = -1.5\text{V}$; $150\text{nm} \times 150\text{nm}$) and **b)** ($V_{\text{Tip}} = -1\text{V}$; $50\text{nm} \times 50\text{nm}$) Images of SL-WSe₂ layer on graphite substrate. Inset: lateral profile corresponding to the dark line in panel **a)** reveals a height of 7Å for SL-WSe₂. **c)** STM image showing 3 types of point defects labelled by black, red and green arrows ($V_{\text{Tip}} = 1\text{V}$; $20\text{nm} \times 20\text{nm}$, Inset: $10\text{nm} \times 10\text{nm}$). **d)** STS spectrum reveals a $2.05 \pm 0.10\text{eV}$ bandgap for SL-WSe₂ ($V_{\text{Tip}} = 1.3\text{V}$, 68.5pA). **e)** Calculated bias-dependent STM images of pristine SL-WSe₂ on graphite (overlaid green spots: Se, blue spots: W.) **f)** PDOS of SL-WSe₂ reveals a 1.5 eV DFT PBE bandgap.

Proposed models for the three defects

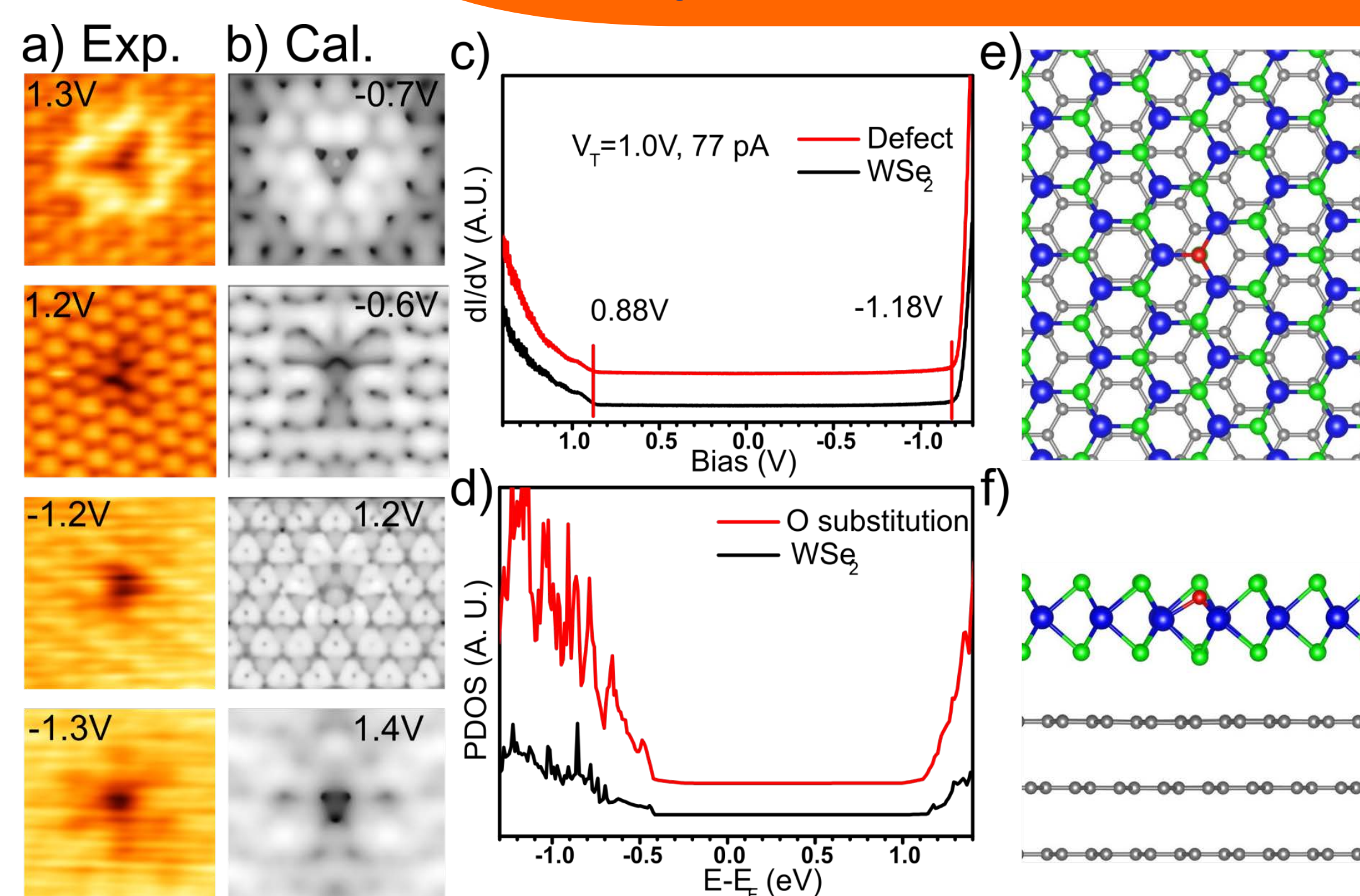


Figure 2. O substitution at Se site (triangular depression, black arrow) in WSe₂. **a)** Bias dependent STM images ($2.5\text{nm} \times 2.5\text{nm}$) of the defect. **b)** Calculated STM images of the O substitution defect. **c)** STS on defect sites and sites far from defects ($> 2\text{nm}$). **d)** PDOS of WSe₂ with and without O substitution. **e)** and **f)**: top and side views of the proposed atomic structure (Red: O, Blue: W, Green: Se and Gray: C.)

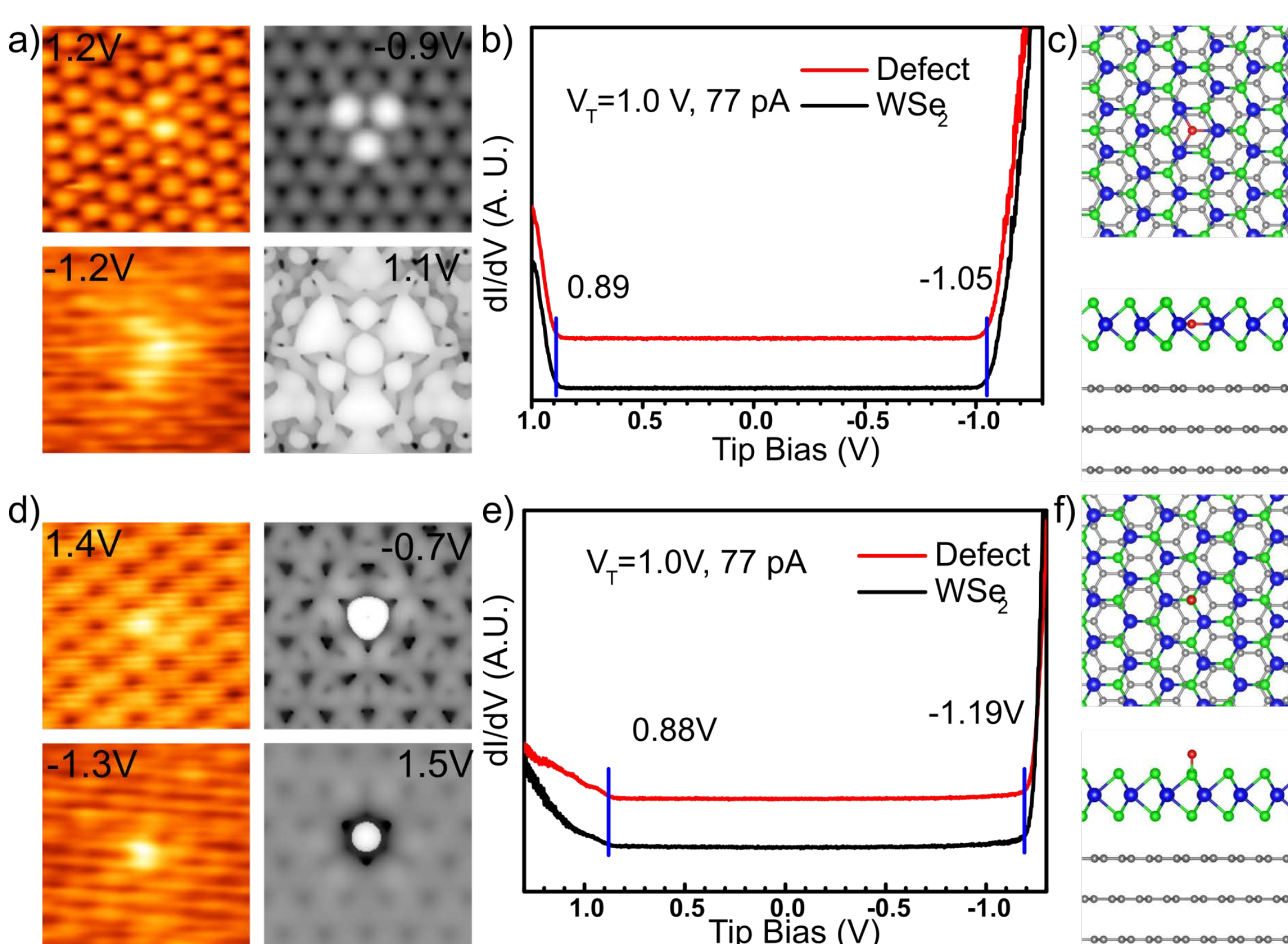


Figure 3. O insertion (bright triangular cluster, red arrow) and O adsorption (bright dot, green arrow) in WSe₂. **a)** and **d)** Bias dependent experimental ($2.5\text{nm} \times 2.5\text{nm}$) and simulated STM images of O insertion and adsorption, respectively. **b)** and **e)** STS on defect sites and sites far from defects ($> 2\text{nm}$) for O insertion and adsorption, respectively. **c)** and **f)** Top and side views of the structure of O insertion in, and O adsorption on WSe₂ layer on graphite substrate, respectively.

Possible formation mechanism

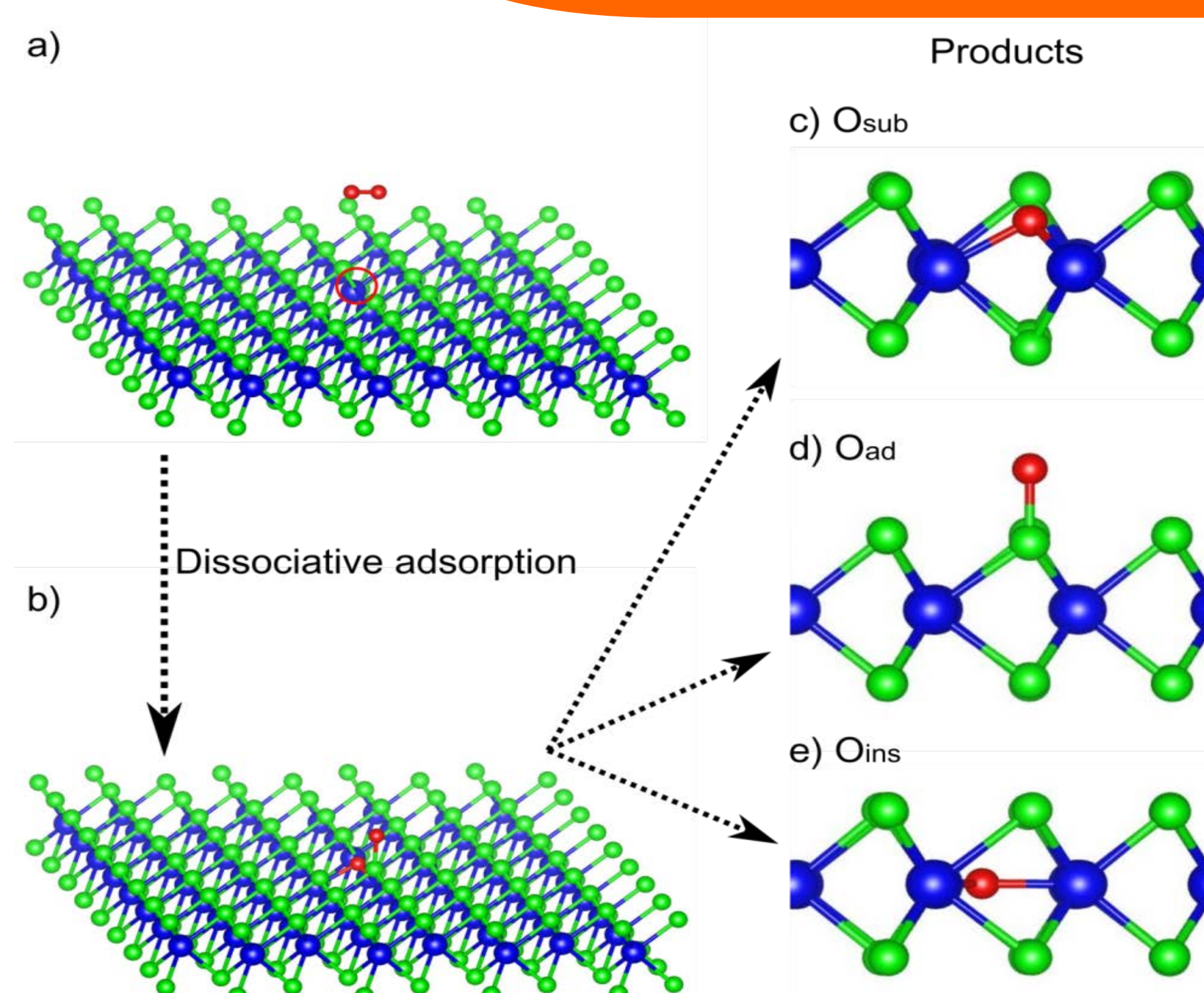


Figure 4. Schematic illustration of the formation of the O-related defects. **a)** O₂ far away from WSe₂ (with Se vacancy). **b)** O₂ dissociative adsorption on WSe₂ (substitution + adsorption). **c-e)** Products. (O substitute Se, another moves to most stable configurations (adsorption + insertion)).